

Paper - VII: Modern Physics - I (New Course)

> Faculty Code: 003 Subject Code: 0492004

Time:  $2\frac{1}{2}$  Hours] [Total Marks: 70

Instructions: (1) All questions are compulsory.

> (2)Figures on the right indicate marks.

- 1 Attempt any **seven** short questions:
- 14 What are the assumptions of Rutherford's model of atom?
  - Write the postulates of Bohr's theory of atomic structure.
  - (3) Draw a well labelled diagram of Spectral lines of H-atom.
  - Define elastic and inelastic collisions. (4)
  - Draw a well labelled diagram of Davison Germer (5)experiment.
  - What is photoelectric effect? Draw a proper diagram (6)showing this effect.
  - (7)How the X-rays are produced?
  - Write the uncertainty principle with necessary mathematical equation for uncertainty in position and
  - (9)Define Bohr Magneton. Write mathematical expression for it.
  - (10) What are different coupling schemes for L and S?
- 2 Write answers of any two:

10

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- Derive an expression for energy of electron (E<sub>x</sub>) in Bohr's orbit.
- Explain the origin of various spectral lines in the (2)emission spectra of H-atom using necessary diagrams.
- Differentiate between Fluorescence and (3) Phosphorescence.
- (4) Describe the Franck - Hertz experiment proving the existence of energy levels in atom.

	(b)	<ul> <li>Write answer of any one:</li> <li>(1) Explain the emission and absorption line spectra.</li> <li>(2) Discuss various models of atom starting from Plum Pudding Model.</li> </ul>	4
3	(a)	Write answers of any <b>two</b> : (1) Write a detailed note on the Angular momenta and Magnetic momenta.	10
		<ul> <li>(2) Explain the concept of Spin-Orbit coupling.</li> <li>(3) Derive Planck's radiation law for the black body.</li> <li>(4) Describe Photoelectric effect using suitable diagram.</li> </ul>	
	(b)	Write answer of any <b>one</b> : (1) Explain the significance of FOUR quantum numbers.	4
		(2) Write a detailed note on Compton effect.	
4	(a)	Write answers of any <b>two</b> : (1) Describe various Selection rules for spectral lines	10
		to be observed.  (2) Describe Raman effect. What are Stokes and Anti Stokes lines?	
		(3) Explain the electronic configuration of an atom and rules for filling up of various orbits.	
		(4) Explain the important facts which are revealed from black body radiation curves.	
	(b)	Write answer of any <b>one</b> :	4
		(1) Describe the significance of Magnetic quantum numbers.	
		(2) What are matter waves? Explain their physical significance.	
5	(a)	Write answers of any two:	10
	,	(1) Describe Stern-Gerlach experiment confirming space quantization.	
		(2) Explain the G.P Thomson's experiment proving the evidence of matter waves.	
		(3) Describe normal and anomalous Zeeman effect.	
	(b)	(4) Explain the Paschen- Back effect. Write answer of any <b>one</b> :	4
	( <i>D)</i>	<ol> <li>Discuss the Stark effect with suitable diagram.</li> <li>Explain the Davison-Germer experiment to verify the wave nature of particles.</li> </ol>	-1